



IDAHO CLEANUP PROJECT

Densification and Shape Change of Calcined High Level Waste during Hot-Isostatic Pressing

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SAFELY PLAN • MOTIVATE • DELIVER

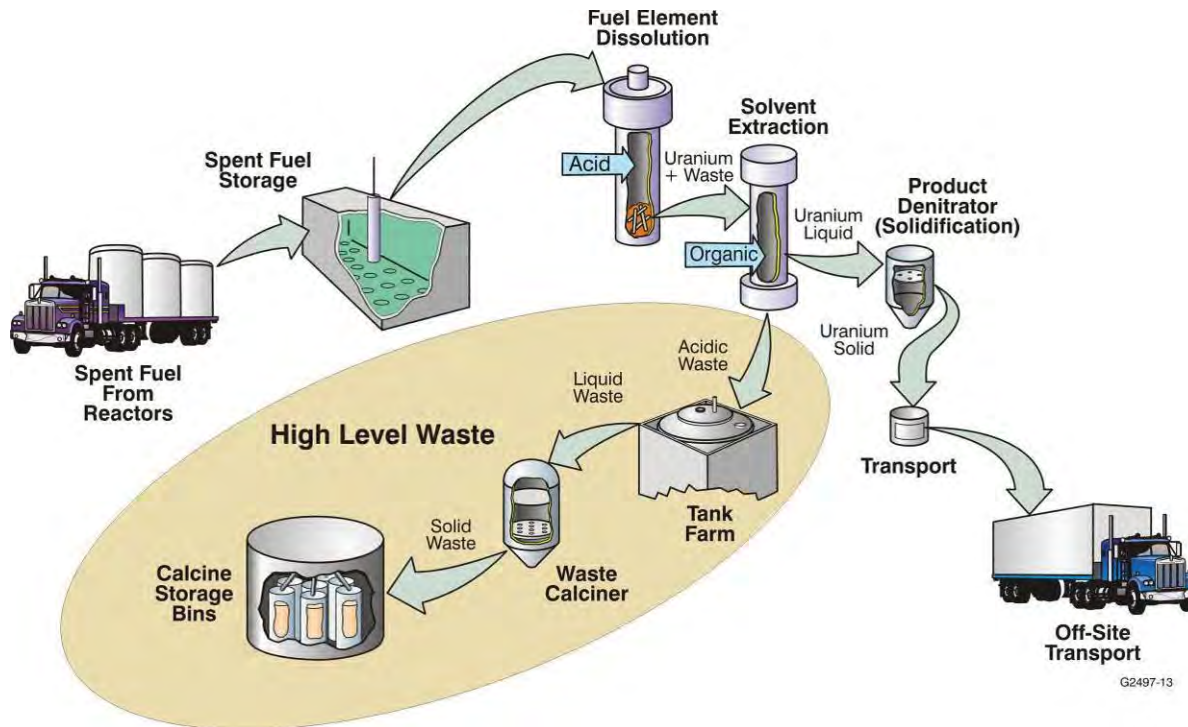
Calcine Densification and Shape Change



- ◆ Troy P. Burnett, P.E.
 - BS Engineering Idaho State University
 - Practicing engineering for 11 years
 - Working on the Calcine Disposition Project for two years
- ◆ Danielle E. Lower
 - BS Mechanical Engineering University of Idaho
 - MS Mechanical Engineering Montana State University
 - Naval Reactors Facility
- ◆ CH2M-WG Idaho (CWI)
 - Contracted to perform work on the Idaho Cleanup Project for the Idaho National Laboratory

Calcine Densification and Shape Change

- ◆ History of Calcine and the Calcine Disposition Project (CDP)
 - The lab received spent nuclear fuel from all over the world
 - Processed fuel from 1953 until 1992
 - Used solvent extraction to separate Uranium
 - Conversion of liquid, high-level radioactive waste to solid by calcination process (high temperature drying)



Calcine Densification and Shape Change

- ◆ What is calcine?
 - Dry granular powder
 - Stored in large bins
 - ◆ 12.2 million pounds of calcine
 - ◆ 4,400 cubic meters
 - Highly radioactive
 - ◆ All handling is done by remote equipment
 - ◆ Must be confined in shielded cells at all times
 - Leachable
- ◆ Purpose of the Calcine Disposition Project
 - Create a stable, long term waste form
 - Process calcine using Hot Isostatic Pressing technology (HIP)



Calcine Densification and Shape Change

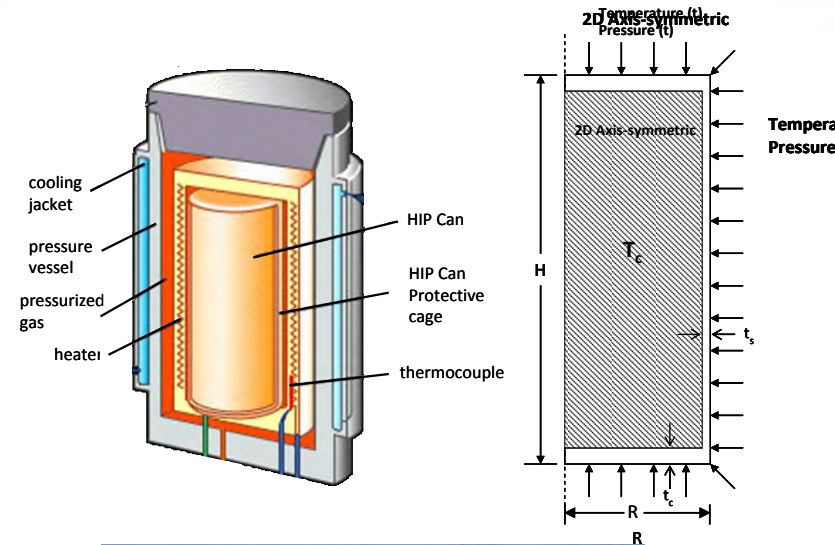
◆ Hot Isostatic Pressing (HIP)

• HIP process

- ◆ A process that is used to reduce the porosity and increases the density of metals and ceramics.
- ◆ Materials are exposed to a high temperature and pressure environment.

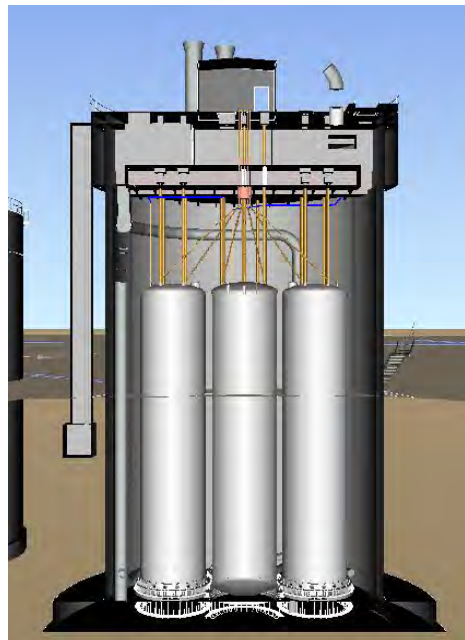
• HIP uses

- ◆ Aerospace industry – Turbine blades
 - ◆ Automotive industry – Turbo chargers, Valves
 - ◆ Medical industry – Prosthetic devices
 - ◆ Petroleum industry – Valve bodies
- ### • HIP use for the Calcine Disposition Project
- ◆ Densification of ceramic powder
 - ◆ Create a stable monolithic waste form

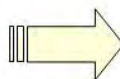


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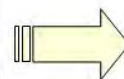
- ◆ The Calcine Disposition Project
 - Retrieve high level waste (HLW) calcine from storage bins
 - Package in Hot Isostatic Press (HIP) can
 - Process filled HIP can using Hot Isostatic Press
 - Package HIP treated can in storage canister



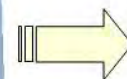
HIP Can



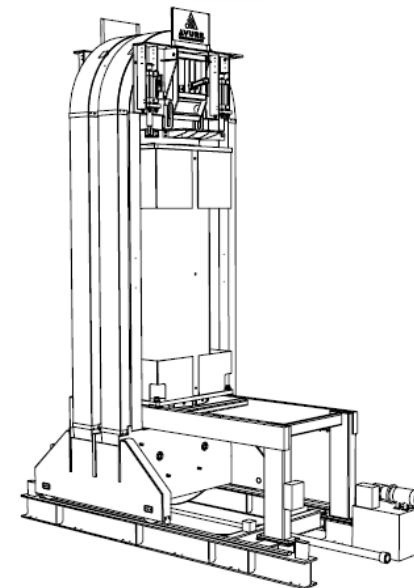
Canister



Cask



Shipping



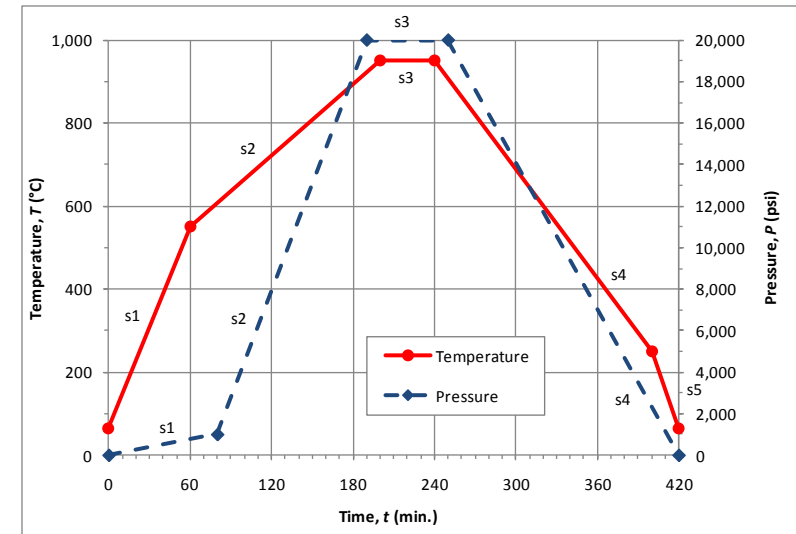
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- ◆ Calcine Disposition Project FEA program
 - Abaqus and COMSOL are being used for analysis of the HIP process as well as the design of process equipment.
 - The focus of this paper is on the analysis of the HIP can
 - The goal for analysis of the HIP can is to accurately predict:
 - ◆ HIP can performance
 - ◆ Final shape of the HIP can
 - ◆ Temperature profile
 - ◆ Densification of calcine
 - The modeling effort will be a multi step process
 - ◆ We are currently in the first stage of the analysis program
 - Will use FEA to streamline the design
 - Large amounts of experimental data will be obtained to validate the FEA
 - The ultimate goal is to incorporate the FEA into the daily facility operations and controls



Calcine Densification and Shape Change

- ◆ Challenges of the analysis
 - This is a transient analysis
 - High temperature – 1000 °C
 - High Pressure – 10 to 20 ksi
 - Large deformation and large strain >30%
 - Multiphysics couplings – heat transfer, solid mechanics, moving mesh, ODE, chemistry, contact, fluid flow
- ◆ Known COMSOL analysis limitations
 - Doesn't handle large strain – large deformation
 - Working with COMSOL to improve in this area
 - Alpha version coming soon
- ◆ Current analysis approach
 - Empirical approach – modification of thermal expansion coefficient
 - Microscopic approach - deformation mechanisms using ODE



Calcine Densification and Shape Change

◆ Empirical Approach – Modified Thermal Expansion Coefficient

• Equations of State

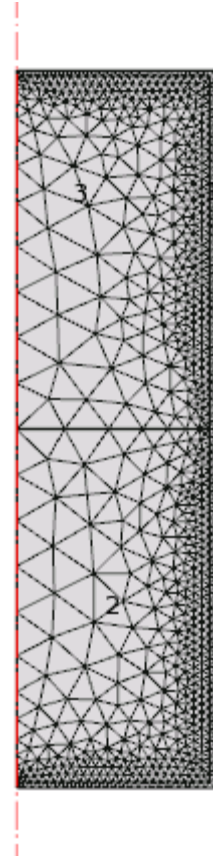
- ◆ Function of temperature and pressure
- ◆ Use equations of state to derive coefficient
- ◆ First-level approximation of volumetric properties
 - Tait equation
 - Spencer-Gilmore Equation

• Thermal expansion coefficient

- ◆ Linear Volume change:

$$\alpha_L = \frac{\left(\frac{\rho_0}{\rho} + \beta\Delta P\right)^{\frac{1}{3}} - 1}{\Delta T}$$

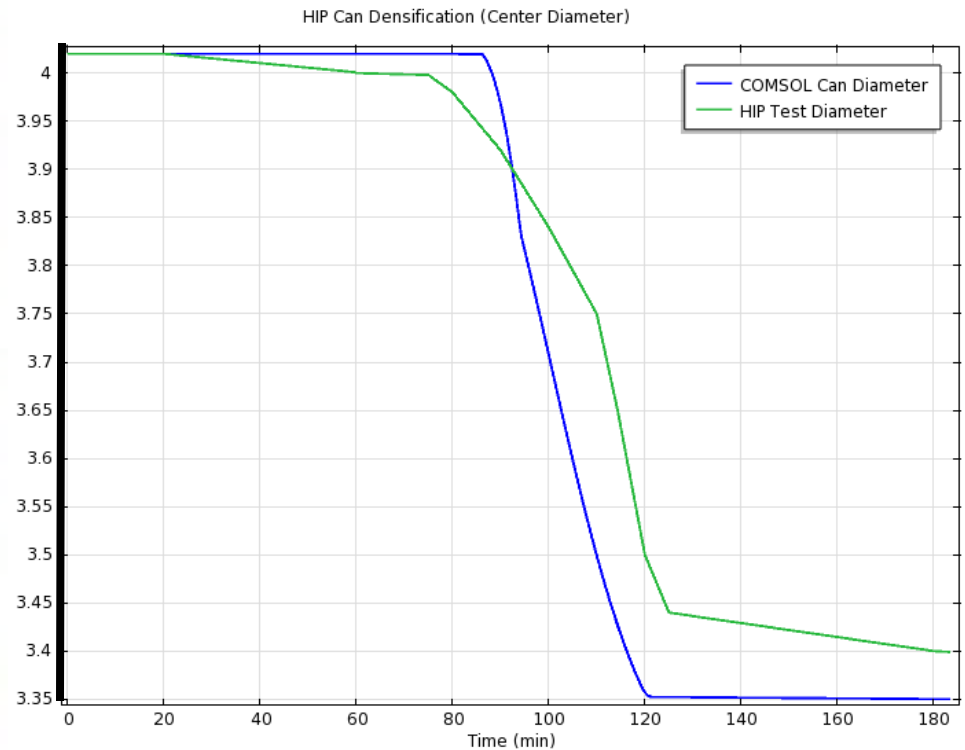
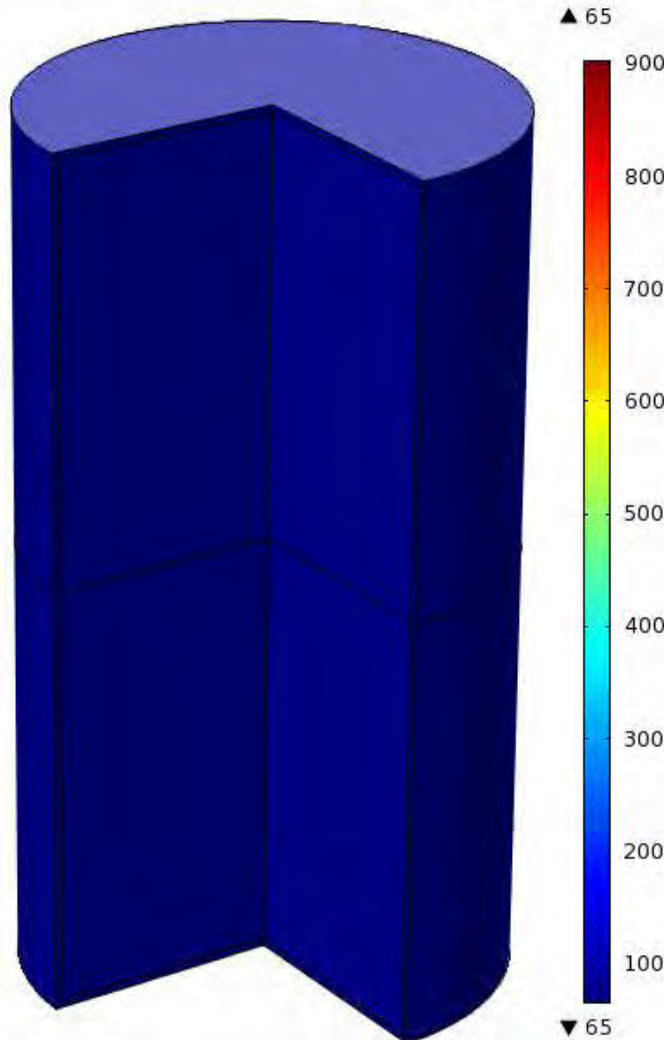
- ◆ ρ and ρ_0 are the final and initial material density
- ◆ β is the material compressibility coefficient
- ◆ Δp is the localized change in pressure
- ◆ ΔT is the localized change in temperature.



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◆ Thermal Expansion Coefficient

Time=0 Surface: Temperature (degC)



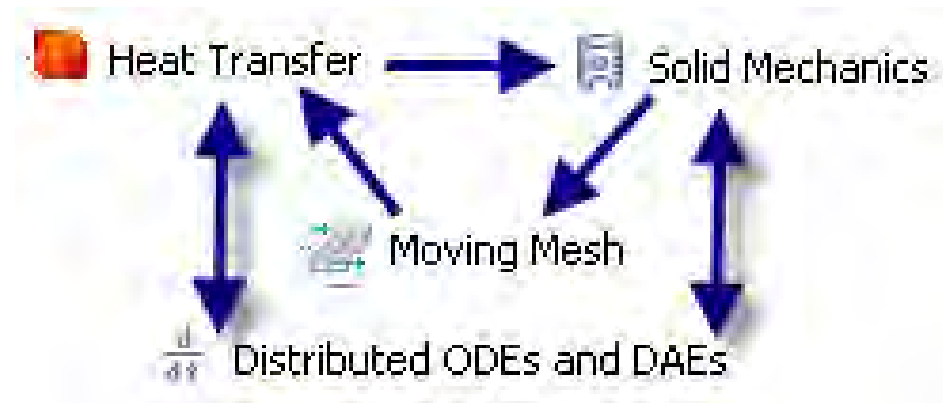
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◆ Microscopic Approach

- M.F. Ashby - Deformation Mechanisms
 - ◆ Plastic Yielding
 - ◆ Power-law creep
 - ◆ Diffusion
- Densification Rate equations
 - ◆ Pressure
 - ◆ Temperature
 - ◆ Relative density
 - ◆ Grain size
- COMSOL Model

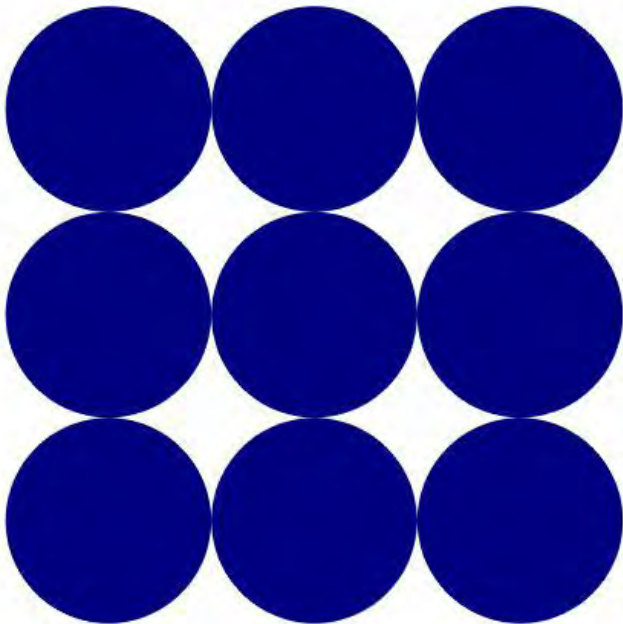
- ◆ Heat transfer
- ◆ Solid Mechanics
- ◆ Moving Mesh
- ◆ Distributed ODE's

$$\frac{d}{dt}D = K_D \cdot f(D)$$



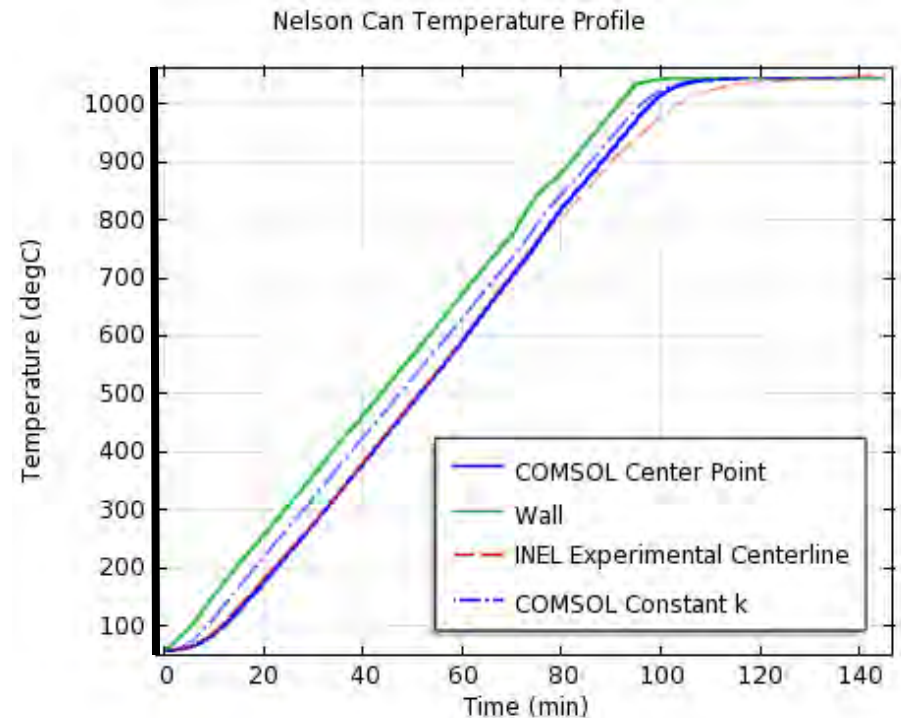
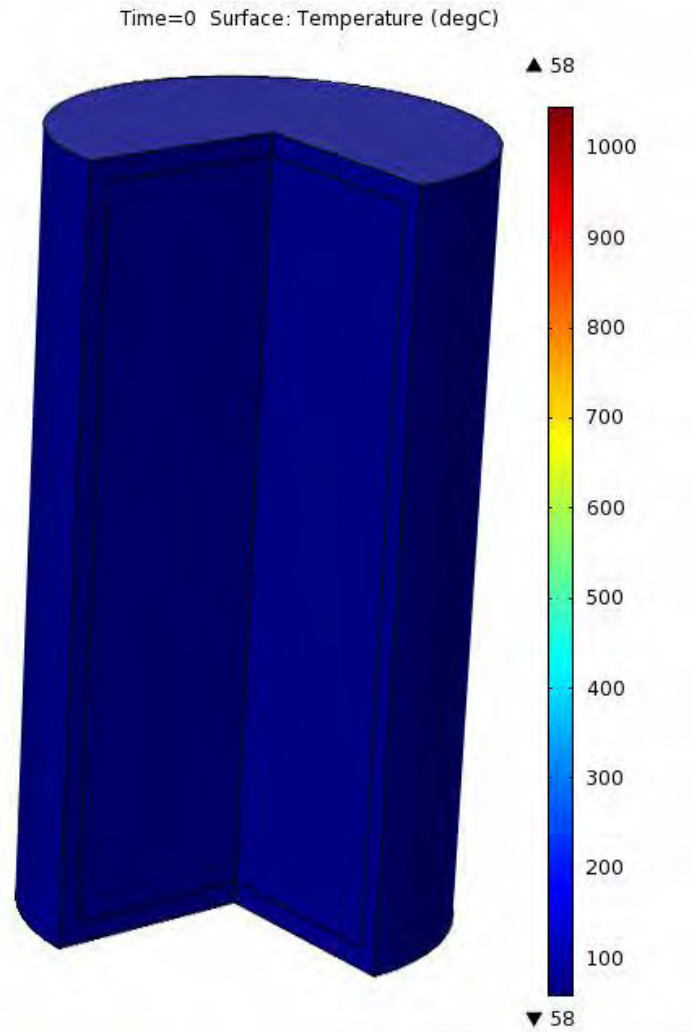
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- ◆ The densification process during HIP
 - Random packing in contact with approximately 7 other particles
 - Particle centers must move together for densification



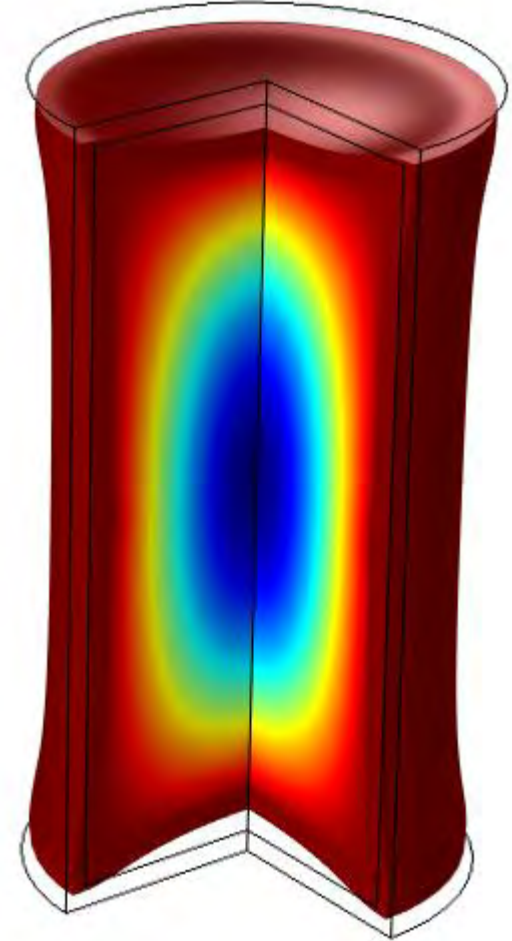
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◆ Deformation Mechanisms



Calcine Densification and Shape Change

- ◆ Conclusion
 - The two methods show good agreement with experimental results in the shape change, densification, and temperature profile
- ◆ Future analysis
 - Incorporate plasticity in the can
 - Incorporate large strain and large deformation
 - Include contact between the can and calcine
 - Include the chemical reactions in the calcine
- ◆ Ultimate goal
 - High fidelity analytical model
 - Use analytic results to validate HIP can performance before the first radioactive can is treated
 - Use analysis in the daily operation and control of the facility



Calcine Densification and Shape Change



◆ Questions?